Supporting information for:

Gadolinium and Platinum in Tandem: Real-time Multi-Modal

Monitoring of Drug Delivery by MRI and Fluorescence Imaging.

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Figure S1. HPLC spectra of the **GdL** and **PtGdL**. Experimental conditions: Agilent ZORBAX SB-C18 Stable Bond Analytical 4.6 X 150 mm 5-micron column, solvent gradient show in table S1.

Time /min	0.05% TFA in water	0.05% TFA in CH ₃ CN /%	Flow rate/	
	/%		mLmin ⁻¹	
0	90	10	1.0	
5	90	10	1.0	
15	60	40	1.0	
20	90	10	1.0	

Table S1. Solvent gradient of the HPLC for characterization of GdL and PtGdL.





Figure S2. ESI-HRMS spectra of GdL and PtGdL. GdL: calcd. for $C_{34}H_{37}GdN_7O_7$ [M + H]⁺ 813.1995, found 813.2005, for $C_{34}H_{36}GdN_7NaO_7$ [M + Na]⁺ 835.1815, found 835.1915. PtGdL: HRMS (+ESI) m/z calcd. for $C_{34}H_{42}ClGdN_9O_7Pt$ [M - NO₃]⁺ 1076.1784, found 1076.1821, for $C_{34}H_{43}ClGdN_9O_8Pt$ [M -NO₃ – Cl + OH]⁺ 1058.2123, found 1058.1654.

Complex	$\lambda_{max}\!/\;nm\;{}^{[a]}$	$\epsilon/\ M^{\text{-1}}\ cm^{\text{-1}\ [a]}$	$q\pm0.2~^{\text{[b]}}$	¢∕ % ^[d]	$r_{l}/mM^{1}s^{1}[^{e]}$
PtGdL	327	22300	1.0 ^[c]	< 0.1	29.4
GdL	324	20000	1.0	3	23.1

Table S2. Photophysical properties and relaxivity of PtGdL and GdL.

a) Absorption coefficient in H₂O, 298 K; b) Coordination number of water molecules, the q value of **GdL** was determined by comparison of the emission lifetime of europium(III) analogue of **GdL**, **EuL** in D₂O and H₂O, which is reported in our previous work. (*Chem. Commun.* 2015, 51, 14022; *J. Chem. Soc., Perkin Trans.* 1999, 2, 493.) c) For PtGdL the q value could not be determined by using the same method with GdL because PtEuL is non-emissive, however PtGdL should have the same hydration number ($q = 1.0 \pm 0.2$) as GdL due to the similarity of structure. d) Fluorescence quantum yield of the ligand emission of the complexes was determined by integrated sphere methods. (*Chem. Rev.* 2010, *110*, 2729.) e) T_1 relativity of the Gd(III) complexes at 10 MHz, 25 °C.

C 1	HeLa		WPMY-1	WPMY-1	
Complex	Dark IC ₅₀	UVA ^[b] IC ₅₀	Dark IC ₅₀	UVA ^[b] IC ₅₀	
PtGdL	23.6 ± 0.8	3.8 ± 0.7	43.9 ± 1.1	$\sim\!\!6.9 \pm 1.1$	
GdL	> 200	> 200	> 200	> 200	
Cisplatin	3.3 ± 0.3	3.4 ± 0.7	4.2 ± 0.6	4.1 ± 0.9	

Table S3. IC_{50} (μ M) values of PtGdL, GdL and cisplatin (as contral) in HeLa and WPMY-1 cells under dark and UVA irradiation.^[a]

a) Incubation time = 24 hour; b) λ_{ex} = 365 nm, light dosage = 1 Jcm⁻². Raw data shows in Figure S3.



Figure S3. Raw data of MTT dark (A and C) and photo (B and D) cytotoxicity in Table S3.



Figure S4. *In vivo* tumor inhibition of **PtGdL** with UVA irradiation. A) The volume, (B) mass and the photograph of the tumor after 14 days treatment of the PBS buffer (control), cisplatin (control) and **PtGdL** + UVA.



Figure S5. The plot of R_1 versus the concentration of PtGdL in aqueous solution. Linear fitting of the data gives a relaxivity of PtGdL = 23.4 mM⁻¹s⁻¹, at 3T and 25°C.

Conc. ^[a] (mol/L)	$T_1 (ms)^{[b]}$	$R_{1} (s^{-1})^{[c]}$
	$(\pm$ SD)	
1.07E-04	343 ± 13	2.91
5.36E-05	713 ± 73	1.40
2.68E-05	1148 ± 25	0.87
1.07E-05	1805 ± 44	0.55
5.36E-06	2047 ± 35	0.49
PBS	2507 ± 45	0.39

Table S4. Raw data of the T₁ relativity of PtGdL at 3T (25°C) in Figure S4.

[a] The concentration of Gd(III) was corrected by ICP-MS. [b] The longitudinal relaxation T_1 is the average of T_1 values of three slices and SD is the standard deviation, for each sample. [c] The relaxation rate, $R_1 = 1/T_1$.





Figure S6. DLS spectrum of **PtGdL** in aqueous solution. The DLS spectrum of **PtGdL** proves that **PtGdL** could form aggregations in aqueous solution.

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